

Cordylite-(Ce)**(Na, Ca)Ba(Ce, La)₂(CO₃)₄F**

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Crystal Data: Hexagonal. *Point Group:* $6/m\ 2/m\ 2/m$. As prismatic or thin to thick tabular crystals with hexagonal outline and dominant $\{10\bar{1}0\}$, $\{40\bar{4}.15\}$, $\{10\bar{1}1\}$, to 13 mm; may have scepterlike terminations with striations $\parallel \{0001\}$; in rosettes, subparallel and spherical aggregates of thin platy crystals.

Physical Properties: *Cleavage:* Good on $\{0001\}$. *Fracture:* Conchoidal to irregular. *Tenacity:* Brittle. Hardness = 4.5 D(meas.) = 4.01–4.44 D(calc.) = 3.97

Optical Properties: Transparent to translucent. *Color:* Colorless, white, wax-yellow, amber-yellow, pale yellow, yellowish green, greenish gray, may be radially zoned, ocher-yellow if altered; colorless to pale yellow in transmitted light. *Streak:* White. *Luster:* Adamantine, vitreous, resinous, oily, pearly on $\{0001\}$; waxy if altered.

Optical Class: Uniaxial (-). *Pleochroism:* Weak; O = greenish yellow; E = brownish yellow. $\omega = 1.764\text{--}1.775$ $\epsilon = 1.576\text{--}1.598$

Cell Data: *Space Group:* $P6_3/mmc$. $a = 5.090\text{--}5.109$ $c = 23.049\text{--}23.289$ $Z = 2$

X-ray Powder Pattern: Mont Saint-Hilaire, Canada.

3.193 (10), 3.510 (9), 4.336 (8), 3.843 (8), 2.550 (8), 2.040 (8), 2.122 (7)

Chemistry: (1)		(2)		(1)		(2)		(1)		(2)	
CO ₂	[25.36]	[24.50]	Pr ₂ O ₃	1.78	1.46	BaO	22.22	20.91			
Ce ₂ O ₃	22.10	23.85	Sm ₂ O ₃		0.21	Na ₂ O	4.73	3.96			
La ₂ O ₃	14.61	14.73	CaO	1.09	0.36	F	3.60	[2.64]			
Nd ₂ O ₃	4.58	4.90	SrO	1.44	0.74	-O = F ₂	1.51	1.11			
							Total	[100.00]	[97.15]		

(1) Narssârssuk, Greenland; by electron microprobe, CO₂ by difference; corresponds to $(\text{Na}_{1.08}\text{Ca}_{0.14})_{\Sigma=1.22}(\text{Ba}_{1.03}\text{Sr}_{0.10})_{\Sigma=1.13}(\text{Ce}_{0.96}\text{La}_{0.64}\text{Nd}_{0.20}\text{Pr}_{0.08})_{\Sigma=1.88}(\text{CO}_3)_{4.10}\text{F}_{1.35}$. (2) Saint-Amable, Canada; by electron microprobe, CO₂ and F calculated from stoichiometry; corresponds to $(\text{Na}_{0.92}\text{Ca}_{0.04})_{\Sigma=0.96}(\text{Ba}_{0.98}\text{Sr}_{0.05})_{\Sigma=1.03}(\text{Ce}_{1.05}\text{La}_{0.65}\text{Nd}_{0.21}\text{Pr}_{0.06}\text{Sm}_{0.01})_{\Sigma=1.98}(\text{CO}_3)_4\text{F}$.

Occurrence: Rare in an alkali pegmatite (Narssârssuk, Greenland); a late-stage mineral in miarolitic vugs and fractures (Saint-Amable, Canada); in pegmatite dikes associated with an intrusive alkalic gabbro-syenite complex (Mont Saint-Hilaire, Canada).

Association: Aegirine, ancylite-(Ce), synchysite-(Ce), parisite-(Ce), neptunite (Narssârssuk, Greenland); analcime, aegirine, albite, pyrophanite (Mont Saint-Hilaire, Canada); parisite, cebaite-(Ce), barite (Bayan Obo deposit, China); vaterite, alstonite, ancylite-(Ce), kukharenkoite-(Ce), mckelveyite-(Y), fluorapatite, barite (Vuoriyarvi complex, Russia).

Distribution: From Narssârssuk, Greenland. In Canada, large crystals at Mont Saint-Hilaire, and from near Saint-Amable, Quebec. In the Khibiny massif, the Vuoriyarvi carbonatite complex, and the Lesnaya Varaka carbonatite, Kola Peninsula, Russia. From the Bayan Obo Fe-Nb-RE deposit, 130 km north of Baotou, Inner Mongolia, China.

Name: From the Greek for *club*, in allusion to its typical clublike crystal habit.

Type Material: University of Copenhagen, Copenhagen, Denmark.

References: (1) Palache, C., H. Berman, and C. Frondel (1951) Dana's system of mineralogy, (7th edition), v. II, 285–287. (2) Chen, T.T. and G.Y. Chao (1975) Cordylite from Mont St. Hilaire, Quebec. *Can. Mineral.*, 13, 93–94. (3) Shen Jinchuan and Mi Jinxiao (1992) A discussion of the composition and structure of cordylite-(Ce). *Acta Petrologica et Mineralogica*, 11(1), 69–74 (in Chinese with English abs.). (4) (1994) *Amer. Mineral.*, 79, 767 (abs. ref. 3). (5) Horváth, L., E. Pfenninger-Horváth, R.A. Gault, and P. Tarassoff (1998) Mineralogy of the Saint-Amable Sill, Varennes and Saint-Amable, Québec. *Mineral. Record*, 29, 83–118, esp. 96. (6) Giester, G., Y. Ni, D. Jarosch, J.M. Hughes, J. Rønsbo, Z. Yang, and J. Zemann (1998) Cordylite-(Ce): a crystal chemical investigation of material from four localities, including type material. *Amer. Mineral.*, 83, 178–184.

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